

(12) **UK Patent Application** (19) **GB** (11) **2 188 630** (13) **A**

(43) Application published 7 Oct 1987

(21) Application No **8707653**

(22) Date of filing **31 Mar 1987**

(30) Priority data

(31) **8608335**

(32) **4 Apr 1986**

(33) **GB**

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(51) INT CL⁴

C07D 211/90 A61K 31/44

(52) Domestic classification (Edition I)

C2C 1531 215 220 22Y 250 251 25Y 29X 29Y 313 31Y 321
32Y 338 364 366 368 36Y 620 624 638 650 652 65X 678
802 80Y AALS
U1S 2415 C2C

(56) Documents cited
None

(58) Field of search
C2C

(54) **Improvements in pharmaceutically acceptable salts**

(57) **Improved pharmaceutical salts of amlodipine, particularly the besylate (benzyl sulphonate) salt, and pharmaceutical compositions thereof.**

This salt finds utility as an anti-ischaemic and anti-hypertensive agent.

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SPECIFICATION

Improvements in pharmaceutically acceptable salts

- 5 The present invention relates to improved pharmaceutical salts of amlodipine and pharmaceutical compositions thereof. 5
- The compound amlodipine (3-ethyl 5-methyl 2-(2-aminoethoxymethyl)-4-(2-chlorophenyl)-1,4-dihydro-6-methylpyridine-3,5-dicarboxylate) is a potent and long acting calcium channel blocker having utility as an anti-ischaemic and anti-hypertensive agent.
- 10 European patent application publication no. 89167 discloses several different pharmaceutically acceptable salt forms of amlodipine. In particular the pharmaceutically acceptable acid addition salts are said to be those formed from acids which form non-toxic acid addition salts containing pharmaceutically acceptable anions such as the hydrochloride, hydrobromide, sulphate, phosphate or acid phosphate, acetate, maleate, fumarate, lactate, tartrate, citrate and gluconate salts. Of these salts the maleate is disclosed as being particularly 15 preferred. 15
- It has now unexpectedly been found that the benzene sulphonate salt (hereinafter referred to as the besylate salt) has a number of advantages over known salts of amlodipine and, additionally has unexpectedly been found to have a unique combination of good formulation properties which make it particularly suitable for the preparation of pharmaceutical formulations of amlodipine.
- 20 Thus according to the present invention there is provided the besylate salt of amlodipine. 20
- In a further aspect the invention provides a pharmaceutical composition of the besylate salt of amlodipine together with a pharmaceutically acceptable diluent or carrier.
- The invention further provides a tablet formulation comprising the besylate salt of amlodipine in admixture with excipients. A preferred formulation includes the besylate salt, a compression aid such as microcrystalline 25 cellulose, an additive to provide sheen to the tablet such as anhydrous dibasic calcium phosphate, a disintegrant such as sodium starch glycollate and a lubricant such as magnesium stearate. 25
- In addition the invention provides a capsule formulation comprising the besylate salt of amlodipine in admixture with excipients. A preferred formulation includes the besylate salt, an inert diluent, a dried disintegrant and a lubricant as described above.
- 30 The invention further provides the besylate salt of amlodipine in sterile aqueous solution for parenteral administration. Preferably such solution contains from 10 to 40% by volume of propylene glycol and preferably also sufficient sodium chloride to avoid haemolysis, e.g. about 1% w/v. 30
- The invention also provides the besylate salt of amlodipine for use in treating ischaemic heart disease, especially angina, or hypertension, in a human being.
- 35 The invention also provides a process for preparing the besylate salt of amlodipine by reacting amlodipine base with a solution of benzenesulphonic acid or its ammonium salt in an inert solvent and recovering the besylate salt of amlodipine. 35
- The preferred inert solvent is industrial methylated spirit.
- Although amlodipine is effective as the free base, in practice it is best administered in the form of a salt of a 40 pharmaceutically acceptable acid. In order to be suitable for this purpose the pharmaceutically acceptable salt must satisfy the following four physicochemical criteria: (1) good solubility; (2) good stability; (3) non-hygroscopicity; (4) processability for tablet formulation, etc. 40
- It has been found that whilst many of the salts outlined above satisfy some of these criteria, none satisfy them all and even the preferred maleate, whilst exhibiting excellent solubility tends to break-down in solution after a few weeks. Consequently a range of pharmaceutically acceptable salts of amlodipine has been made 45 and evaluated using these criteria: 45
1. Generally, it is known in the art that a good aqueous solubility is necessary for good bioavailability. Usually a solubility of greater than 1 mg ml^{-1} at pH 1-7.5 is sought although higher solubilities are required to formulate injections. In addition salts which provide solutions having a pH close to that of blood (7.4) are 50 preferred because they are readily biocompatible and can easily be buffered to the required pH range without altering their solubility. 50
- As can be seen from the following comparative data the besylate salt of amlodipine exhibits good solubility characteristics, compared with other salts.

TABLE 1

Salt	solubility mg ml ⁻¹	pH at saturation	
5			5
Benzene sulphonate (besylate)	4.6	6.6	
Toluene sulphonate (tosylate)	0.9	5.9	
Methane sulphonate (mesylate)	25	3.1	
Succinate	4.4	4.9	
10 Salicylate	1.0	7.0	10
Maleate	4.5	4.8	
Acetate	50	6.6	
Hydrochloride	50	3.5	
15	2. Good stability in the solid state is very important for tablets and capsules, whilst good stability in solution is required for an aqueous injection.		15
	In order to screen for chemical stability, each of the salts was blended in a powder vehicle and formed into tablets or capsules. In the case of tablets the vehicle comprised microcrystalline cellulose in 50:50 combination with anhydrous dibasic calcium phosphate. In the case of capsules the vehicle comprised mannitol in 4:1 combination with dried maize starch. These were then stored in sealed vials at 50 and 75°C for up to three weeks. The drug and any breakdown products were extracted with methanol:chloroform (50:50) and separated on silica tlc plates using a variety of solvent systems.		20
	The results were compared and the salts ranked according to the number and amount of breakdown products produced.		
25	By comparing the results the following rank order emerges with besylate being the most stable salt and hydrochloride the least stable.		25
Salt	Stability		
Besylate	most stable		
Mesylate			
30	Tosylate		30
Succinate			
Salicylate	↓		
Maleate			
Acetate			
35	Hydrochloride	unstable	35
	3. In order to provide stable formulations it is desirable to have a non-hygroscopic salt. In the solid state where drug content is high, absorbed films of moisture can act as a vector for hydrolysis and chemical breakdown. It is the hygroscopic nature of a drug or its salt which contributes to the free moisture which is normally responsible for instability.		40
	Only the maleate, tosylate and besylate salts do not pick up any moisture when exposed to 75% relative humidity at 37°C for 24 hours. Even when exposed to 95% relative humidity at 30°C for 3 days both the besylate and maleate remain anhydrous whilst the tosylate formed the dihydrate salt. Therefore the besylate salt can be considered to be non-hygroscopic and thus provides stable formulations while minimising the risk of intrinsic chemical breakdown.		45
	4. The final characteristic of an acceptable salt to be considered is the processability, i.e. the compression properties and also the ability not to stick or adhere to the tablet making machinery.		
	For high dose formulations, good compressability is very important to make elegant tablets. With lower dose tablets the need for good compressability can be eliminated to a certain extent by the use of suitable diluting excipients called compression aids. Microcrystalline cellulose is a commonly used compression aid. However whatever the dose the adhesion of the drug to the punches of the tablet machine must be avoided. When drug accumulates on the punch surfaces this causes the tablet surface to become pitted and therefore unacceptable. Also sticking of the drug in this way results in high ejection forces when removing the tablet from the machine. In practice it is possible to reduce sticking by wet-massing, careful selection of excipients and the use of high levels of anti-adherents, e.g. magnesium stearate. However selection of a salt with good anti-adhesion properties minimises these problems.		50
	In order to compare the stickiness of the various salts of amlodipine the following procedure was carried out using conventional tablet making machinery: fifty tablets containing calcium sulphate dihydrate, microcrystalline cellulose and amlodipine besylate were made (47.5:47.5:5), the material sticking to the tablet punch was then extracted using methanol and the amount measured spectrometrically. This procedure was then repeated for runs of 100, 150, 200, 250 and 300 tablets. After each run the amount of material sticking to the table punch was measured after extraction with methanol. The values are plotted and an average value calculated from the slope of the line produced.		60
65	This same procedure was then repeated for each of the salts of amlodipine. The amount of amlodipine measured as sticking to the tablet punch is shown in Table 2 for each salt and relative to the maleate salt.		65

TABLE 2

Salt	Stickiness	
	$\mu\text{g Amlodipine cm}^{-2}$ tablet ⁻¹	Relative to maleate
5 Mesylate	1.16	58%
Besylate	1.17	59
10 Tosylate	1.95	98
Maleate	1.98	100
Free base	2.02	102
Succinate	2.39	121
Hydrochloride	2.51	127
15 Salicylate	2.85	144

Clearly the besylate has superior anti-adhesion properties to the maleate. Whilst the mesylate also shows good processability it tends to be isolated as the anhydride but this equilibrates to the monohydrate leading to variable composition after manufacture which makes it unacceptable for use in tablets.

20 Thus the besylate salt of amlodipine shows a unique combination of good solubility, good stability, non-hygroscopicity and good processability which makes it outstandingly suitable for the preparation of pharmaceutical formulations of amlodipine.

In order that the present invention be more readily understood, reference is now made to the following Examples.

25

EXAMPLE 1

Preparation of besylate salt of amlodipine

Amlodipine base (65.6 g, 0.161 mols) was slurried in industrial methylated spirit (326.4 ml) and cooled to 5°C. Benzenesulphonic acid (26.2 g, 0.168 mols) was dissolved in industrial methylated spirit (65.6 ml) at 5°C and added to the slurry of the base. The resulting slurry was then granulated, filtered and washed with 2 volumes of industrial methylated spirit (65.6 ml). The damp solid was slurried at 5°C for 1 hr in industrial methylated spirit (327.6 ml), filtered, washed with 2 volumes of industrial methylated spirit (65.6 ml) and dried under vacuum at 55°C for 24 hours. A yield of 76.5 g (83.8%) was obtained with the following analysis.

35 Melting Point 201.0°C

Analysis %	C	H	N
Calc.	55.07	5.51	4.94
Found	54.91	5.46	4.93

40

EXAMPLE 2

Formulation of tablets containing besylate salt of amlodipine

Amlodipine besylate was blended with sodium starch glycollate and anhydrous dibasic calcium phosphate for 5 minutes. This mixture was then sieved, reblended and sieved again followed by blending with microcrystalline cellulose. The resultant mixture was then sieved again and blended for a further 10 minutes. Finally magnesium stearate was added and the whole mixture blended for 5 minutes. The blend was then pressed into tablets using conventional tablet making machinery.

This method was used to make tablets containing different concentrations of the amlodipine besylate salt as shown in Table 3.

50

TABLE 3

Tablet compositions

Besylate salt (mg)	Microcrystalline cellulose (mg)	Anhydrous dibasic calcium phosphate (mg)	Sodium starch glycollate (mg)	Magnesium stearate (mg)
55 1.736	63.514	31.750	2.00	1.00
3.472	62.028	31.500	2.00	1.00
6.944	124.056	63.000	4.00	2.00
60 13.889	248.111	126.000	8.00	4.00

EXAMPLE 3

Formulation of capsules containing besylate salt of amlodipine

Microcrystalline cellulose and dried maize starch were pre blended. The besylate salt of amlodipine was then mixed with some of this preblend and then sieved. The remainder of the preblend was then added and

mixed for 10 minutes. This was then sieved again and mixed for a further 5 minutes.

This method was used to make mixtures containing different concentrations of the amlodipine besylate salt as shown in Table 4 and the mixtures were then filled into capsules of appropriate size.

5 TABLE 4

Capsule compositions

	<i>Besylate salt (mg)</i>	<i>Microcrystalline cellulose (mg)</i>	<i>Dried maize starch (mg)</i>	<i>Magnesium stearate (mg)</i>	<i>Total Capsule weight (mg)</i>	
10	1.736	38.014	10.00	0.0250	50	
	3.472	76.028	20.00	0.500	100	
	6.944	72.556	20.00	0.500	100	
15	13.889	145.111	40.00	1.00	200	15

EXAMPLE 4

Formulation of sterile aqueous solution of besylate salt of amlodipine

Sodium chloride was dissolved in water for injection and propylene glycol was mixed with this solution. The besylate salt of amlodipine was added and, when it had dissolved, further water for injection was added to adjust the volume to give the desired concentration of amlodipine (1 mg/ml). The solution was then filtered through a sterilising filter and filled into suitable sterile containers, e.g. ampoules, for use in parenteral, e.g. intravenous, administration.

This method was used to prepare the formulations shown in Table 5.

25

TABLE 5

Sterile Aqueous Solutions

	(1)	(2)	
Besylate salt of amlodipine	1.389 g	1.389 g	
30 Sodium chloride	9.000 g	9.000 g	30
Propylene glycol	200.000 g	400.000 g	
Water for injection	to 1 litre	to 1 litre	

EXAMPLE 5

35 *Alternative preparation of besylate salt of amlodipine*

Ammonium benzenesulphonate (0.943 g) was added to a slurry of amlodipine base (2 g) in industrial methylated spirit (10 ml) and the resulting solution was heated at reflux for 10 minutes. The reaction mixture was cooled and granulated at 5°C for 1 hour. The amlodipine benzenesulphonate was filtered, washed with industrial methylated spirit (2×2 ml) and dried in vacuum.

40

Yield 1.9 g (70% of theory).
Mpt.: 201.0°C.

Analysis %:-

45	Found:	C,54.98; H,5.46; N,4.90;	45
	Calculated for:	C,55.07; H,5.51; N,4.95.	

CLAIMS

- 50 1. The besylate salt of amlodipine.
2. A pharmaceutical composition comprising the besylate salt of amlodipine as in claim 1 together with a pharmaceutically acceptable diluent or carrier.
3. A tablet formulation comprising the besylate salt of amlodipine as claimed in claim 1 in admixture with excipients.
- 55 4. A tablet formulation as claimed in claim 3 wherein the excipients comprise a compression aid, an additive to provide sheen to the tablet, a disintegrant and a lubricant.
5. A tablet formulation as claimed in claim 4 wherein the excipients comprise microcrystalline cellulose, anhydrous dibasic calcium phosphate, sodium starch glycolate and magnesium stearate.
6. A capsule formulation comprising the besylate salt of amlodipine as claimed in claim 1 in admixture with excipients.
- 60 7. A capsule formulation as claimed in claim 6 wherein the excipients comprise an inert diluent, a dried disintegrant and a lubricant.
8. A capsule formulation as claimed in claim 1 wherein the excipients comprise microcrystalline cellulose, dried maize starch and magnesium stearate.
- 65 9. A sterile aqueous solution comprising the besylate salt of amlodipine for parenteral administration.

10. A sterile aqueous solution as claimed in claim 9 comprising from 10 to 40% w/v of propylene glycol.
11. A sterile aqueous solution as claimed in claim 9 or claim 10 comprising about 1% w/v sodium chloride.
12. The besylate salt of amlodipine for use in treating heart disease or hypertension.

Printed for Her Majesty's Stationery Office by Croydon Printing Company (UK) Ltd, 8/87, 08991685.
Published by The Patent Office, 25 Southampton Buildings, London WC2A 1AY, from which copies may be obtained.

L1 ANSWER 1 OF 1 WPIDS COPYRIGHT 2005 THE THOMSON CORP on STN
 AN 1983-809527 [45]
 DNC C1983-108063
 TI Sultamicillin (chlorobenzene sulphonate salts - are stable antibacterials especially useful for paediatric dosage forms.
 DC B02
 IN BARTH, W E: JASYS, V J
 PA (PFIZ) PFIZER INC
 CYC 27
 PI EP 92968 A 19831102 (198345)* EN 26

R: AT BE CH DE FR GB IT LI LU NL SE
 AU 8313871 A 19831117 (198401)
 NO 8301430 A 19831114 (198401)
 JP 58201792 A 19831124 (198402)
 DK 8301017 A 19831212 (198405)
 FI 8301392 A 19831230 (198407)
 US 4432987 A 19840221 (198410)
 HU 29113 T 19840130 (198411)
 DD 209632 A 19840516 (198437)
 PT 76587 A 19840824 (198437)
 CS 8302817 A 19840831 (198447)
 ES 8406075 A 19841016 (198450)
 ZA 8302845 A 19841022 (198507)
 CA 1187484 A 19850521 (198525)
 SU 1138030 A 19850130 (198534)
 EP 92968 B 19860312 (198611) EN
 R: AT BE CH DE FR GB IT LI LU NL SE
 DE 3362502 G 19860417 (198617)
 IL 68453 A 19860831 (198642)
 KR 8601279 B 19860905 (198702)
 JP 01019395 B 19890411 (198918)

ADT EP 92968 A EP 1983-302238 19830420; JP 58201792 A JP 1983-71315 19830422;
 US 4432987 A US 1982-371156 19820423; ZA 8302845 A ZA 1983-2845 19830422;
 SU 1138030 A SU 1983-3578255 19830419

PRAI US 1982-371156 19820423

REP FR 2493322; GB 2084572

IC A61K031-43; C07C000-00; C07D499-22; C07D519-00

AB EP 92968 A UPAB: 19930925

4-X-Benzenesulphonic acid addition salts (I) of sultamicillin
 (6-(D-(2-amino-2-phenyl acetamido)) penicillanate) and their hydrated
 forms are new. (X is H or Cl).

(I) especially the crystalline dihydrates, have excellent pharmacokinetic
 properties, near optimal solubility in aqueous systems and improved stability
 in bulk and aqueous suspensions c.f. sultamicillin free base. (I) are especially
 useful for formulating paediatric dosage forms (i.e. oral dosage forms).

(I) allows rapid absorption from the gastro-intestinal tract.

(I) are antibacterials liberating ampicillin and sulbactam on in vivo
 hydrolysis. Daily dose is 20-100 mg/kg p.o. and 10-100 mg/kg parenterally.

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MC CPI: B02-P; B02-P02

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